

Cubanite & associated sulfides in CI chondrites & comet 81P/Wild 2: Implications for aqueous processing

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Workshop on Water in Asteroids and Meteorites
Comets
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Characteristics of Stardust Sulfides
Fe, Ni, Cu, Zn, Mn-bearing
Mineralogy, Structure, Associations

Modified by capture into aerogel

Unmodified by capture into aerogel

Nebular Formation
via gas-solid processes

Parent Body Precipitation

Aqueous Modification

on Wild 2

on other body

Implications for the genesis of Wild 2 sulfides, source material(s), and mixing in the early Solar System as indicated by compositions, structures, and uniformity/heterogeneity of phases.

Sulfide Populations

		Comet Wild 2	CI Chondrites
Minerals	Cubanite CuFe_2S_3	✓*	✓*
	Pyrrhotite $(\text{Fe}, \text{Ni})_{1-x}\text{S}$	✓*	✓*
	Pentlandite $(\text{Fe}, \text{Ni})_9\text{S}_8$	✓*	✓
	Sphalerite $(\text{Fe}, \text{Zn})\text{S}$	✓*	?
Assemblages	Cubanite & Pyrrhotite	?	✓*
	Pyrrhotite & Pentlandite	✓*	✓
	Pyrrhotite & Sphalerite	✓*	?

* Berger et al. (2011); Pentlandite: e.g., Bullock et al. (2005)

CuFe_2S_3

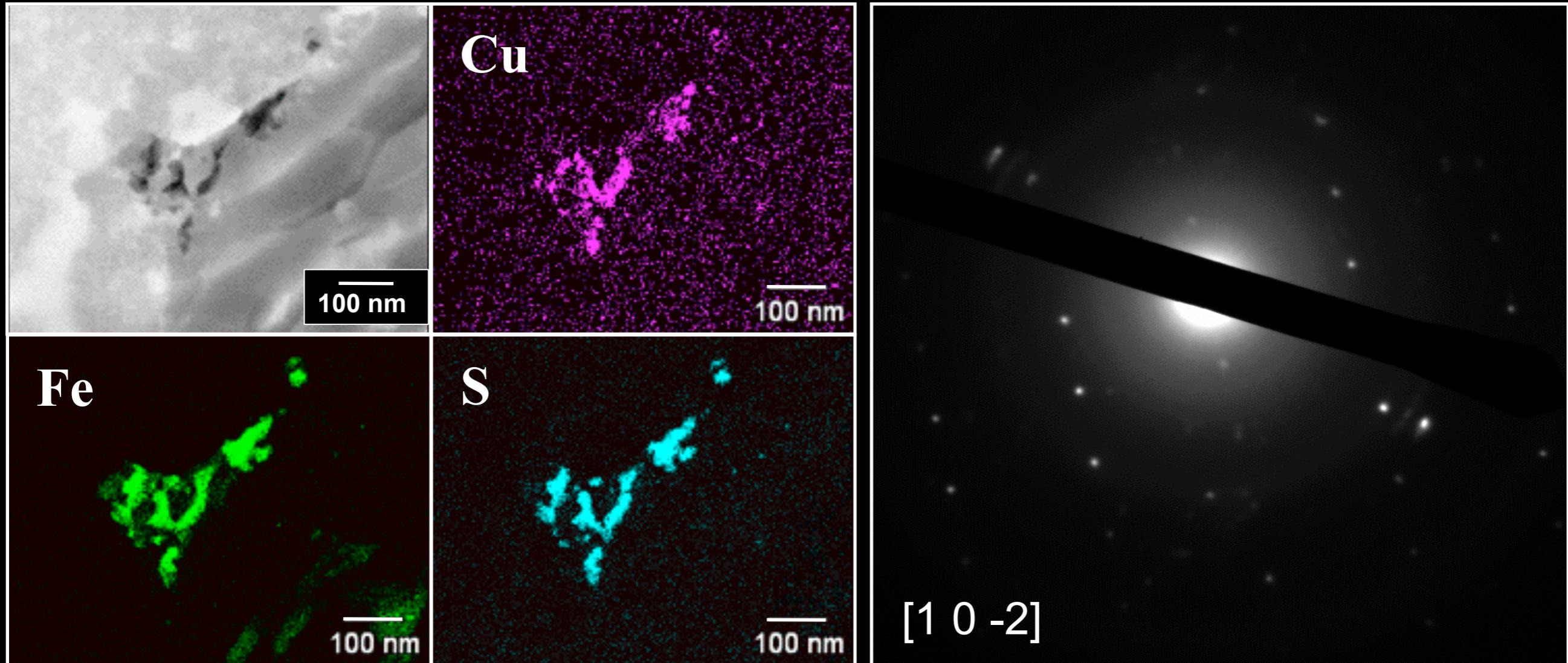
Cubanite undergoes an irreversible phase transition at 210°C



Caye *et al.* (2000), Miyamoto *et al.* (1980), Pruseth *et al.* (1999), Putnis *et al.* (1977)

Stardust Cubanite

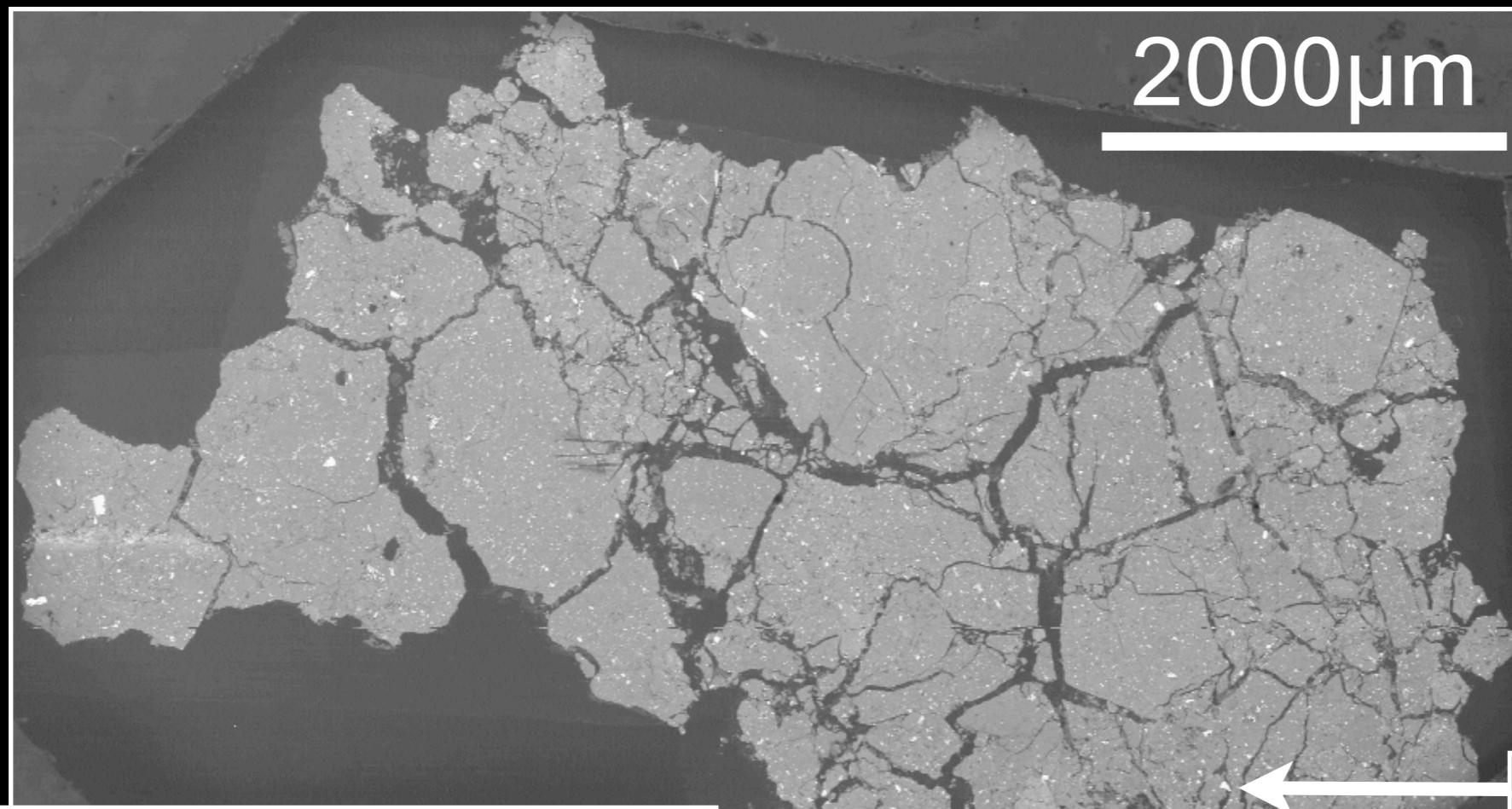
C2054-5-26-I-16



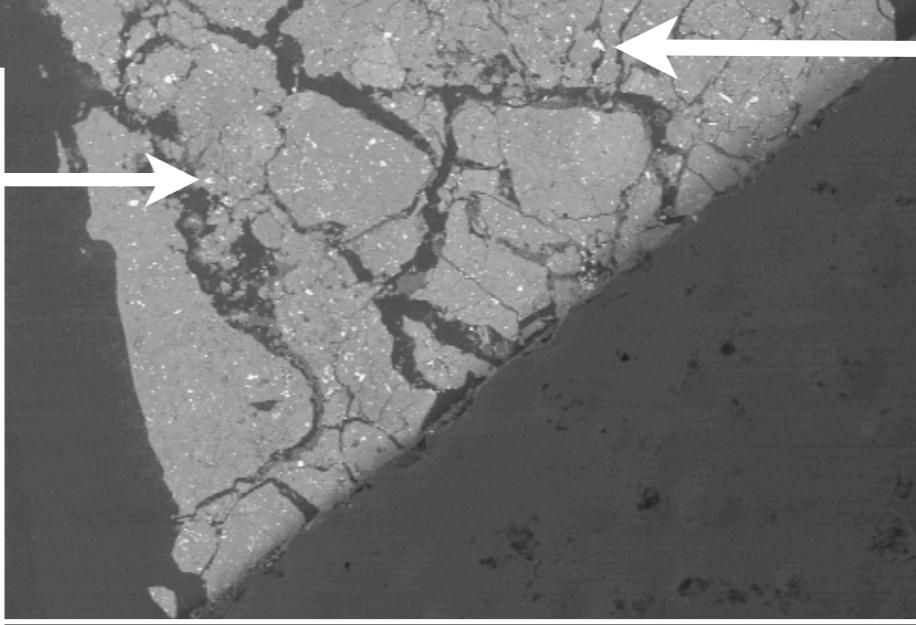
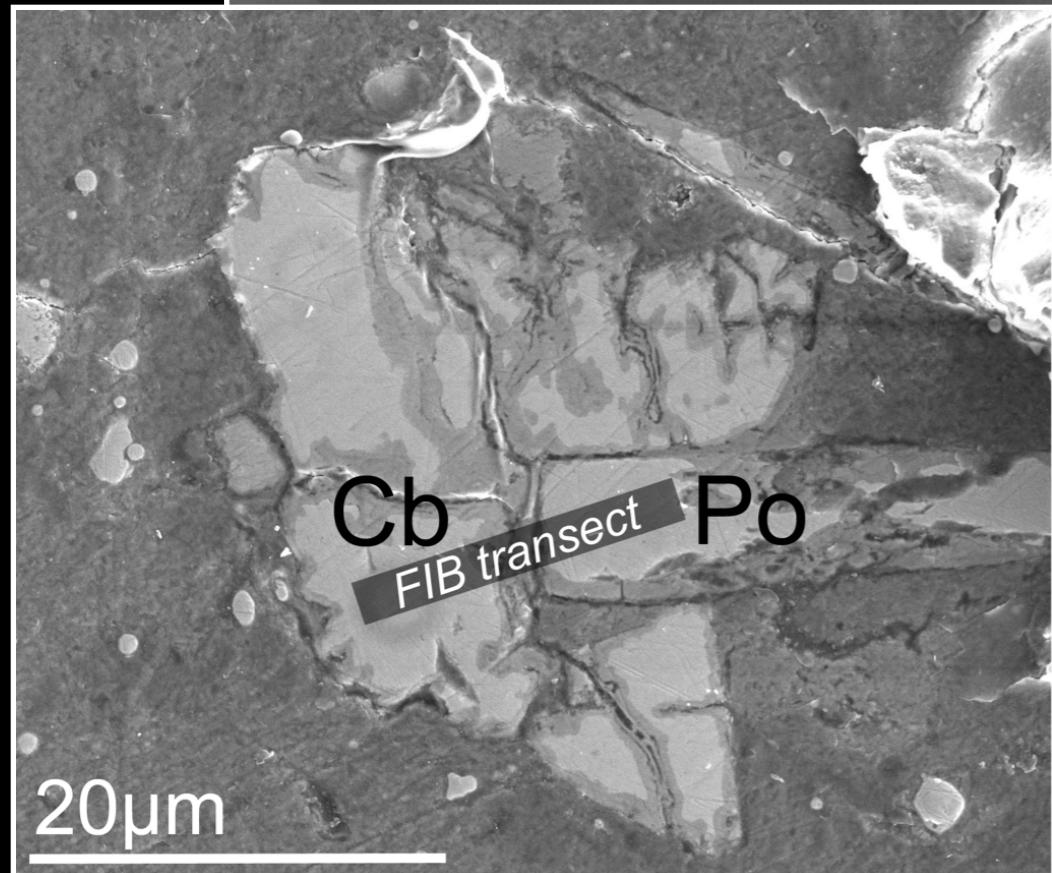
15.3 at.% Cu
35.0 at.% Fe
49.7 at.% S

*Low-temperature
orthorhombic cubanite never
experienced $T > 210^{\circ}\text{C}$*

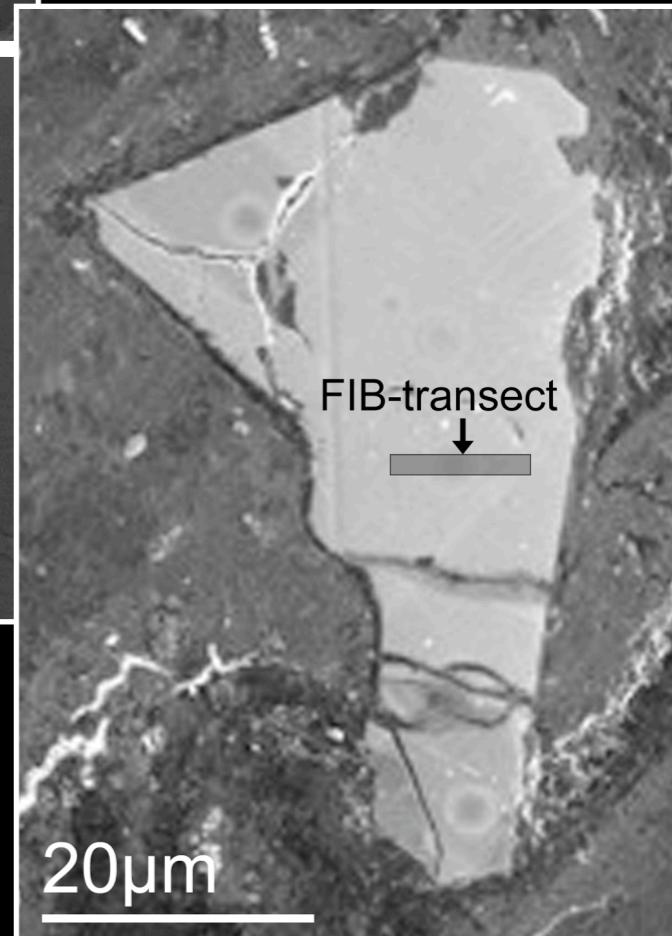
Cl-chondrite Sulfides



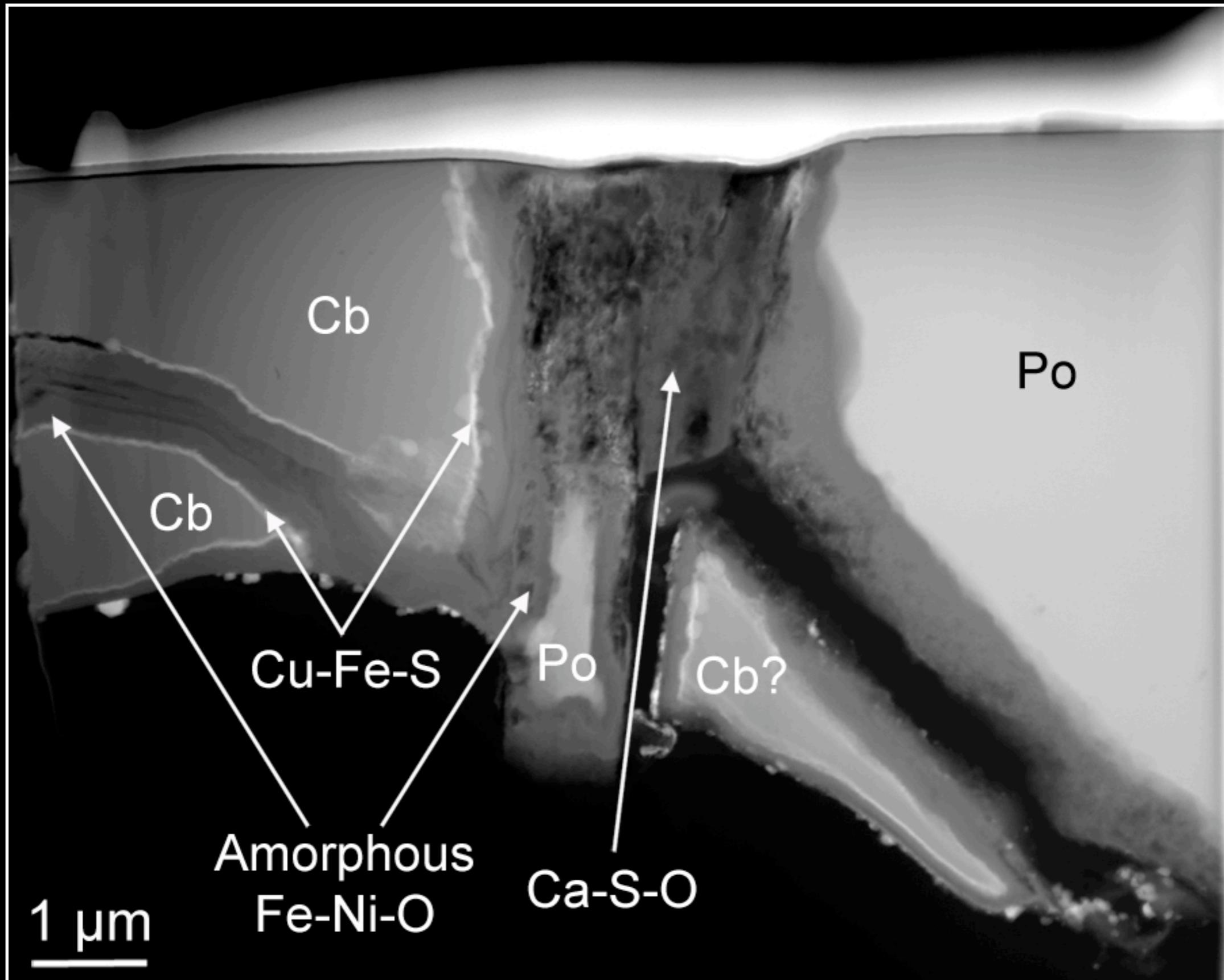
Cubanite



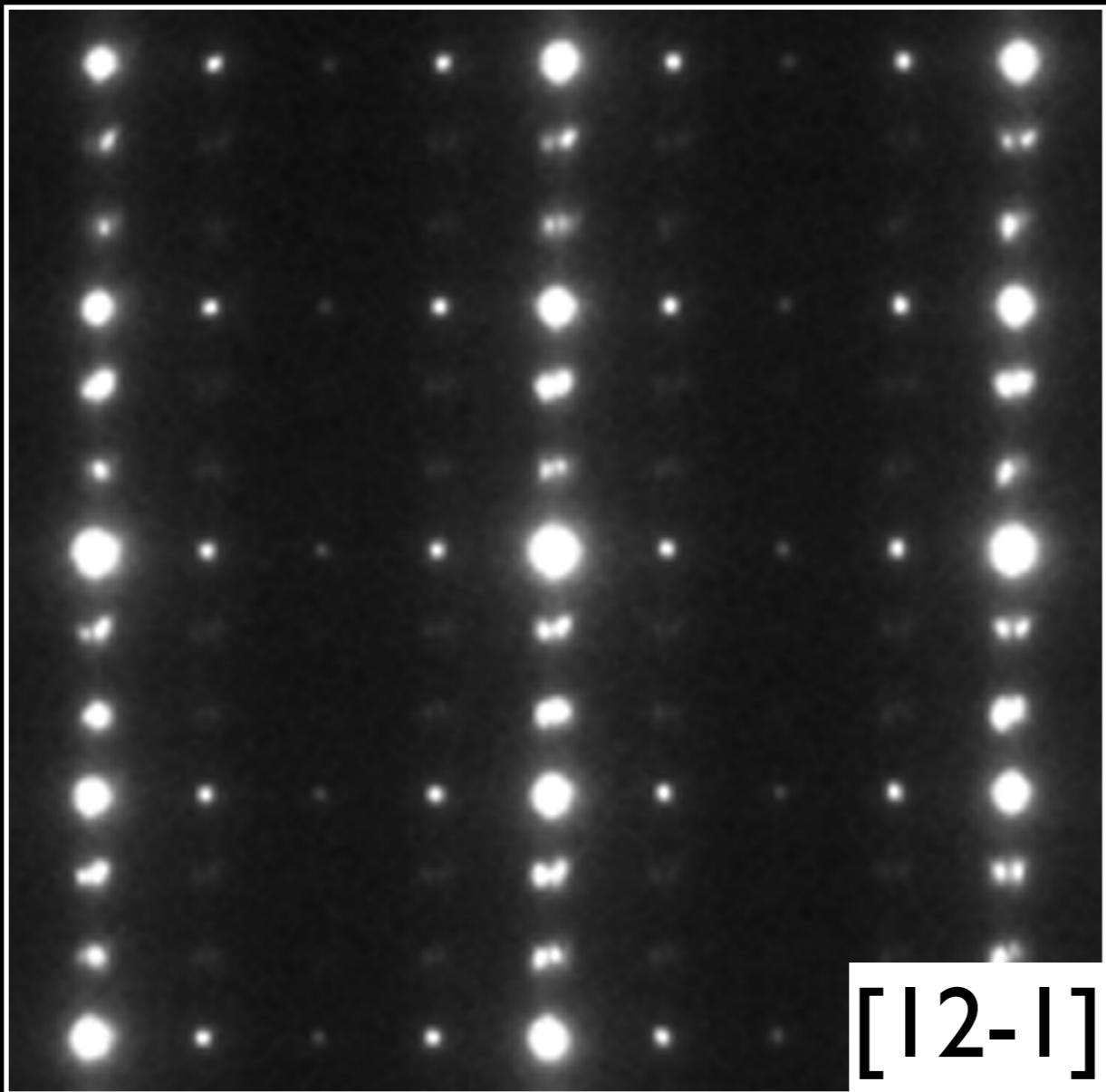
Cubanite
& Pyrrhotite



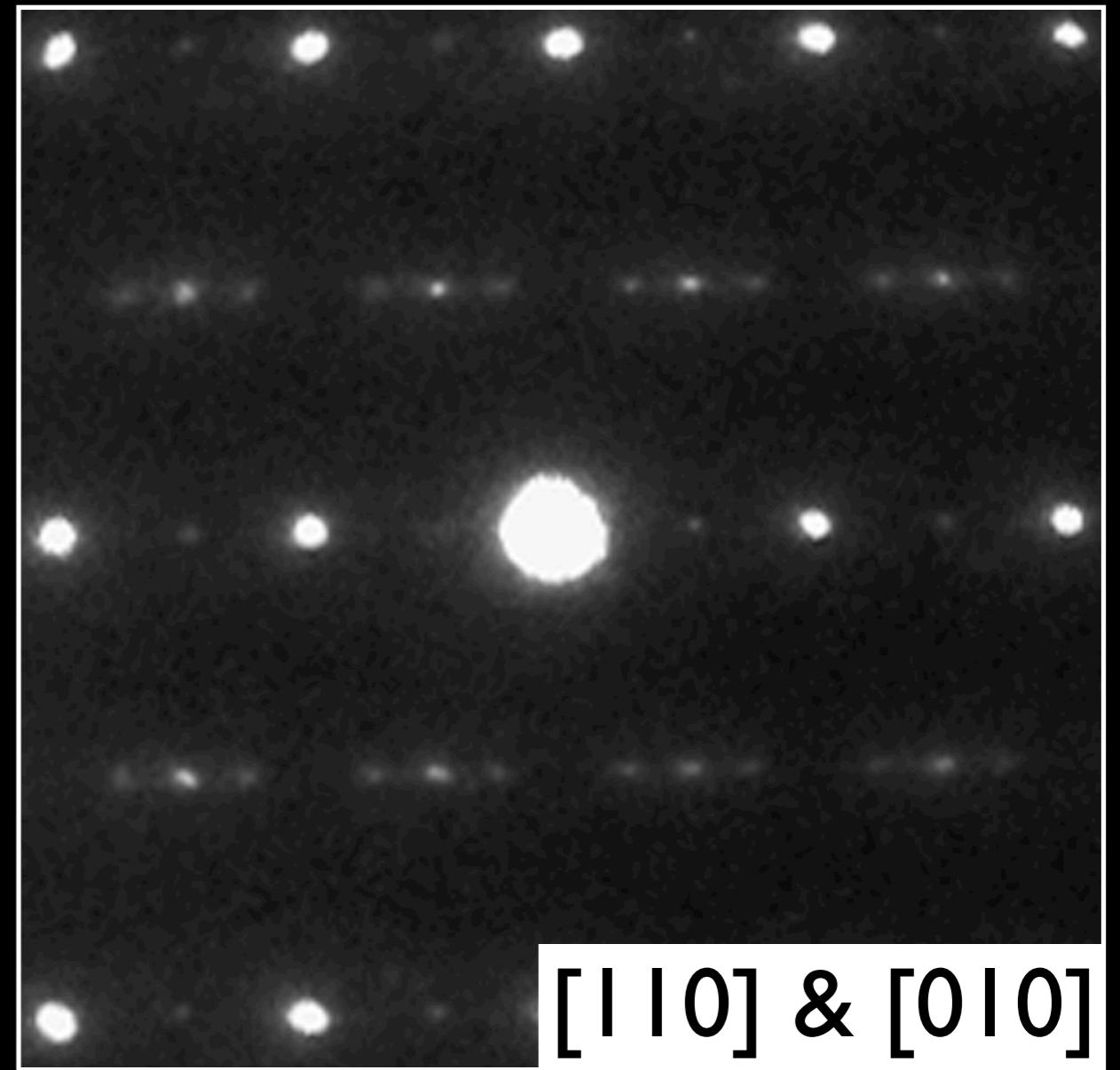
CI-chondrite Cubanite & Pyrrhotite



Cl-chondrite Cubanite & Pyrrhotite



cubanite
 CuFe_2S_3



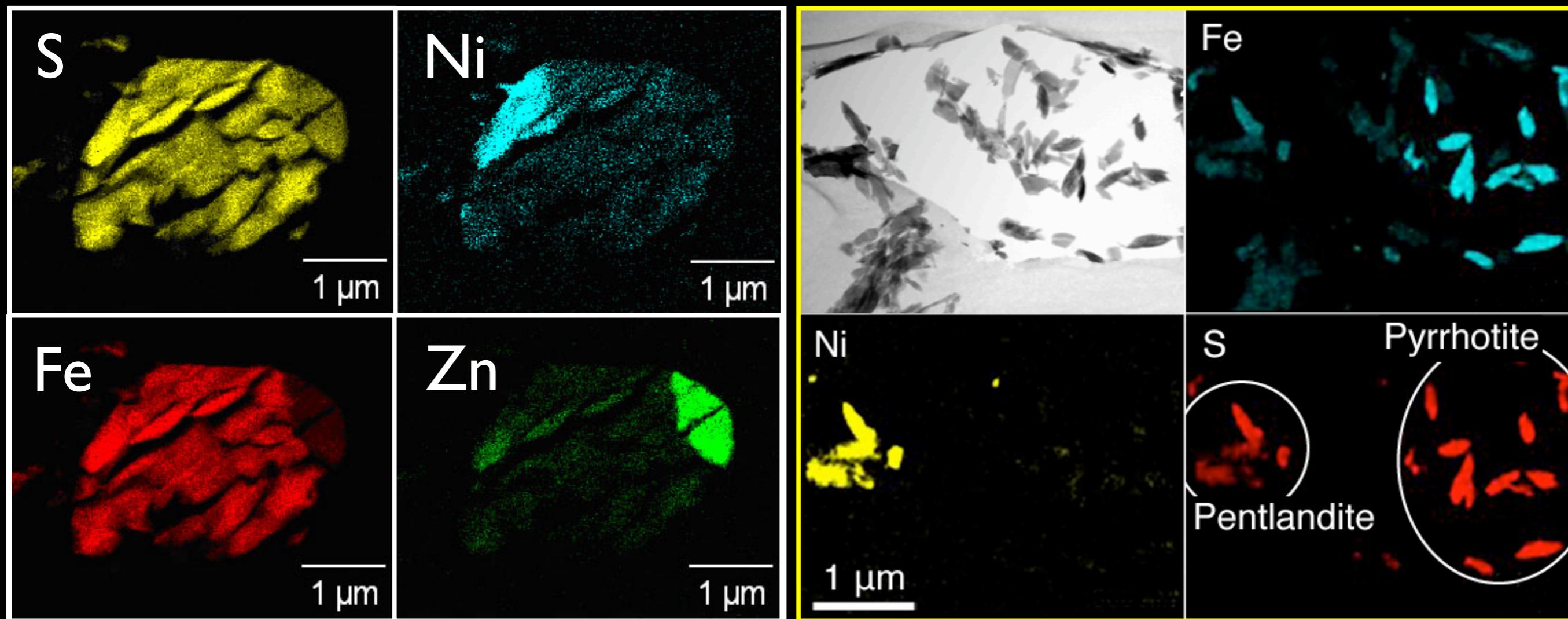
4C monoclinic pyrrhotite
 $(\text{Fe},\text{Ni})_7\text{S}_8$

Combination of pyrrhotite & cubanite indicates $T < 200^\circ\text{C}$

Stardust Fe-Ni-Sulfides

Pyrrhotite
FC6-0-10-0-85

Pyrrhotite & Pentlandite
C2054-5-27-1-11



4C monoclinic pyrrhotite indicates $T < 250^\circ\text{C}$

4C monoclinic pyrrhotite & pentlandite are a stable assemblage on the 100-135°C Fe-Ni-S ternary diagram

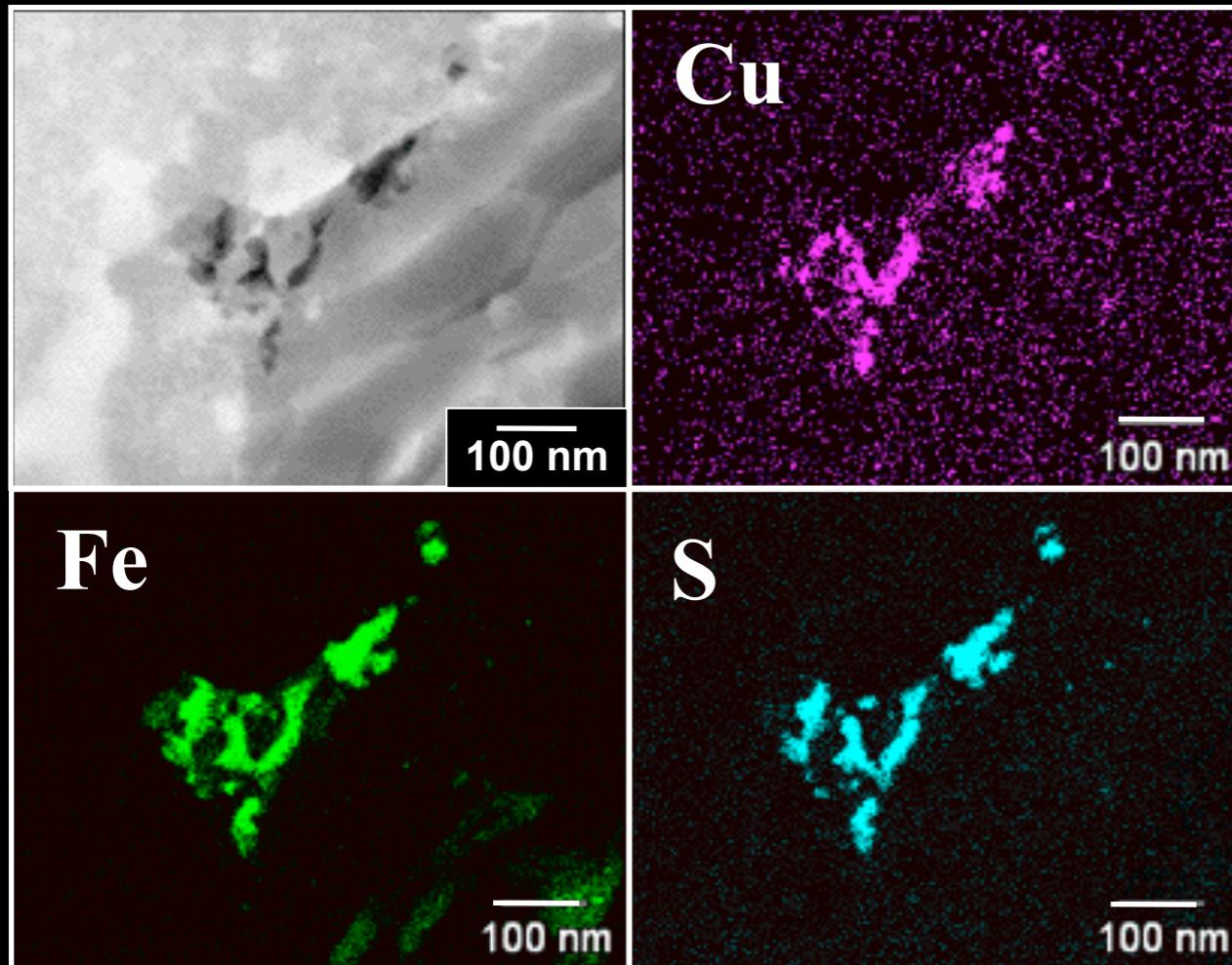
Mineral Constraints

	Wild 2	CI Chondrites		
Minerals	Cubanite CuFe_2S_3	✓	✓	$T_{\max} = 210^\circ\text{C}$
	4C Pyrrhotite Fe_7S_8	✓	✓	$T < 250^\circ\text{C}$
	Pentlandite $(\text{Fe},\text{Ni})_9\text{S}_8$	✓	✓	$T < 610^\circ\text{C}$
	Sphalerite $(\text{Fe},\text{Zn})\text{S}$	✓	?	$T < 1200^\circ\text{C}$
Assemblages	Cubanite & Pyrrhotite	?	✓	$T < 200^\circ\text{C}$ Fe-Cu-S
	4C Pyrrhotite & Pentlandite	✓	✓	$T < 250^\circ\text{C}$ Fe-Ni-S
	4C Pyrrhotite & Sphalerite	✓	?	250°C Fe-Zn-S

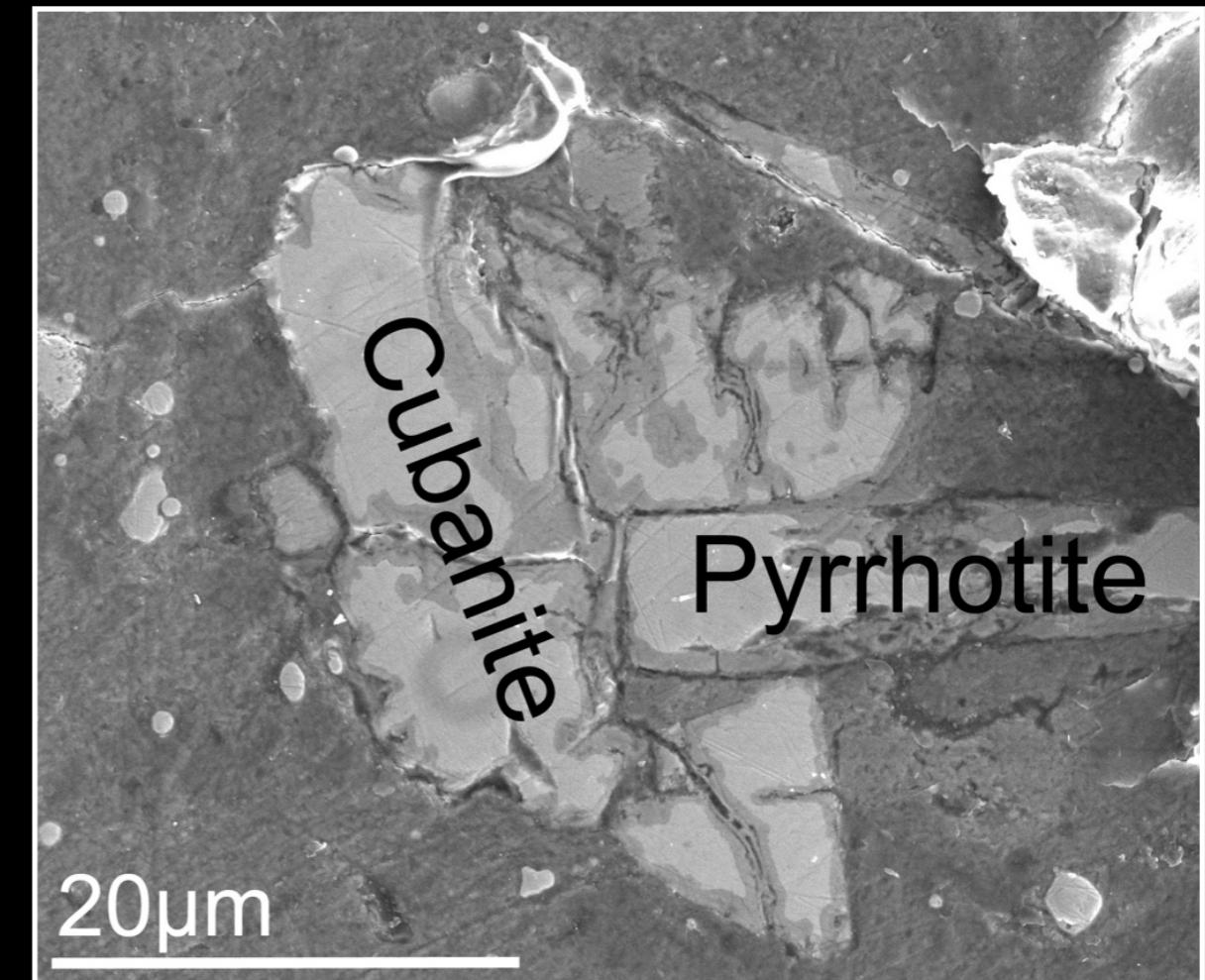
1: Caye et al. (2000), Miyamoto et al. (1980), Pruseth et al. (1999), Putnis et al. (1977); 2: Wang et al. (2006); 3: Fleet (2006); 4: Ihto (1999);
 5: Vaughan & Craig (1997), Yund & Kullerud (1966); 6: Naldrett (1989); 7: Kojima & Sugaki (1984), Scott & Kissin (1973), Vaughan & Craig (1997).

In situ hydrothermal recrystallization

Stardust Cubanite



Orgueil Sulfides



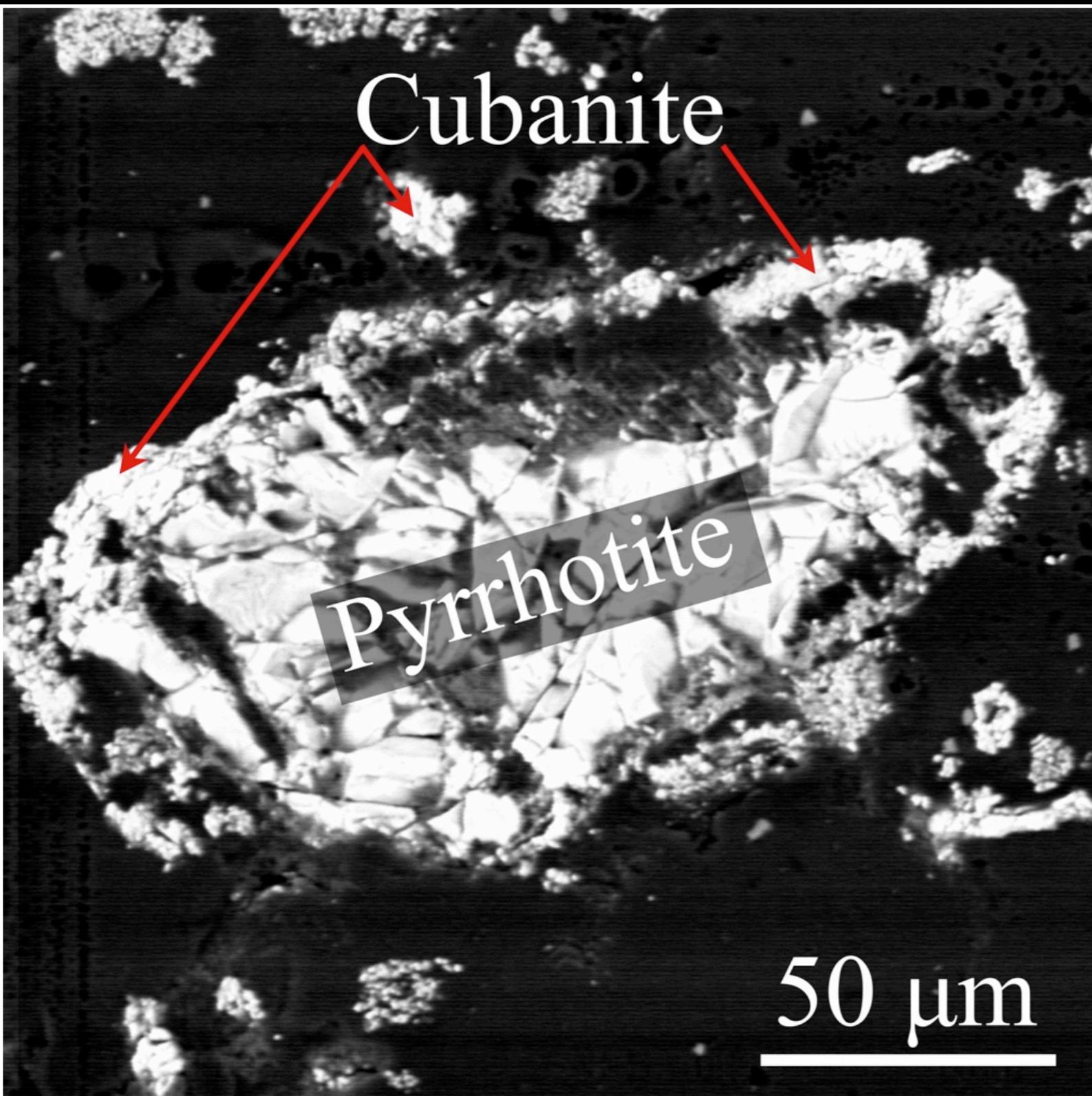
150-200°C

starting pH = 9, 9.5, & 10

starting fO₂ at 1M buffer (solids and powders)

Cu, Fe, S, FeS, CuS

Experimentally produced Cubanite/pyrrhotite assemblage



Fluid conditions consistent with those predicted for the Cl-chondrite parent body:

- ★ 200°C
- ★ $\text{FeS} + \text{CuS}$
- ★ Starting pH = 9.01
- ★ Started in the presence of Fe and Fe_3O_4 powders

Cubanite overgrowths on pyrrhotite are directly comparable to Orgueil sulfide assemblages

Genesis of Wild-2 Cubanite

Scenario 1

Nebular formation of cubanite

Delivery to the comet-forming region

Scenario 2

Nebular formation of troilite

Parent-body aqueous alteration

Delivery to comet-forming region

Scenario 3

Nebular formation of troilite

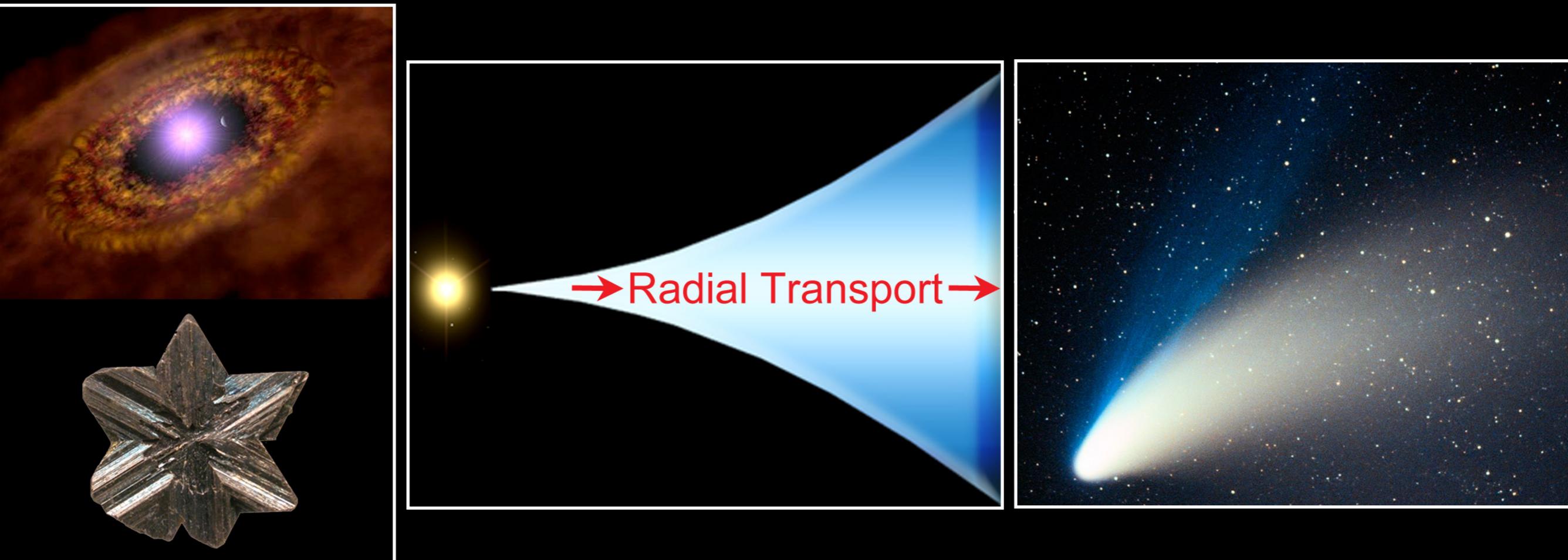
Delivery to comet-forming region

Cometary-body aqueous alteration

Genesis of Wild-2 Cubanite

Option 1

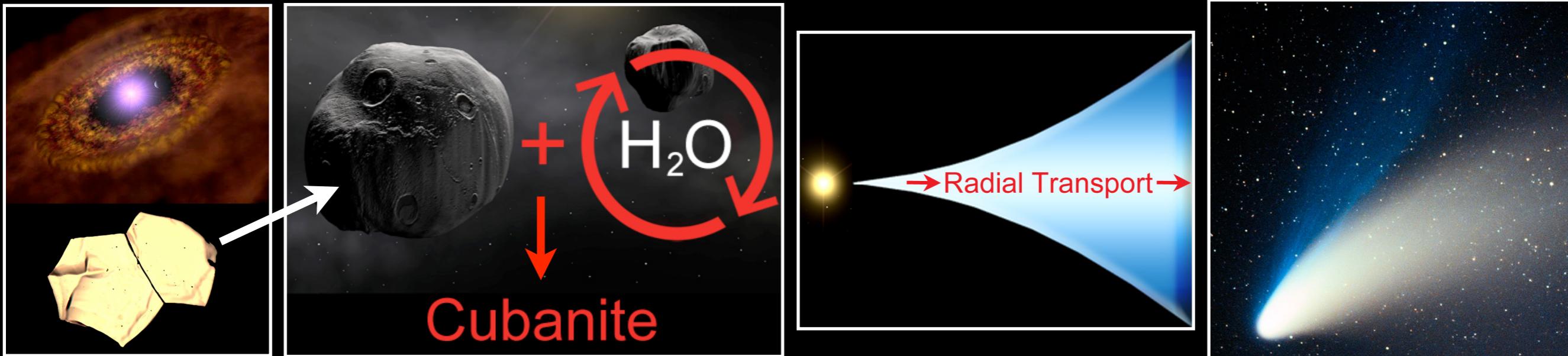
- ★ Nebular formation of cubanite
- ★ Delivery of cubanite to comet-forming region



Genesis of Wild-2 Cubanite

Option 2

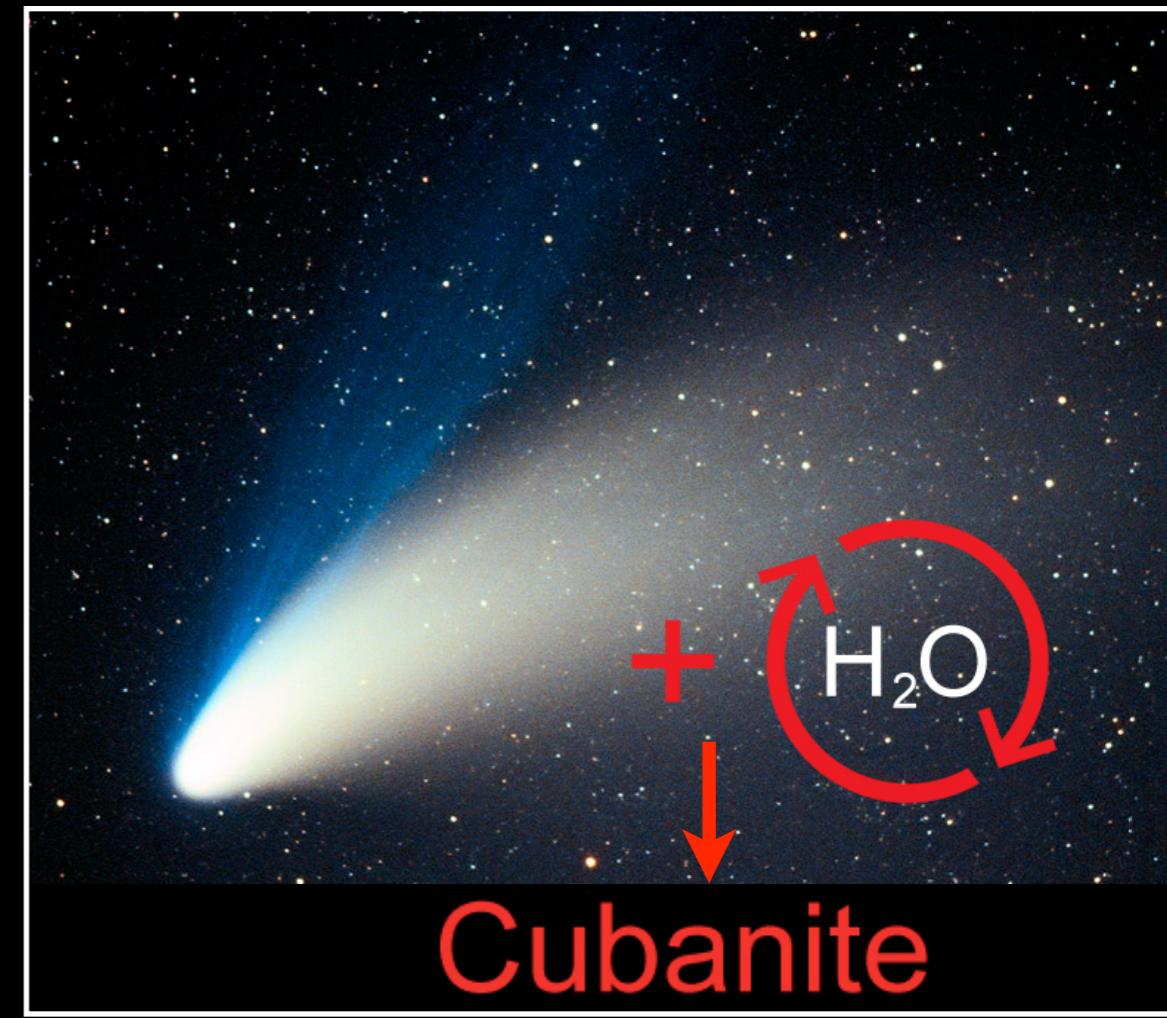
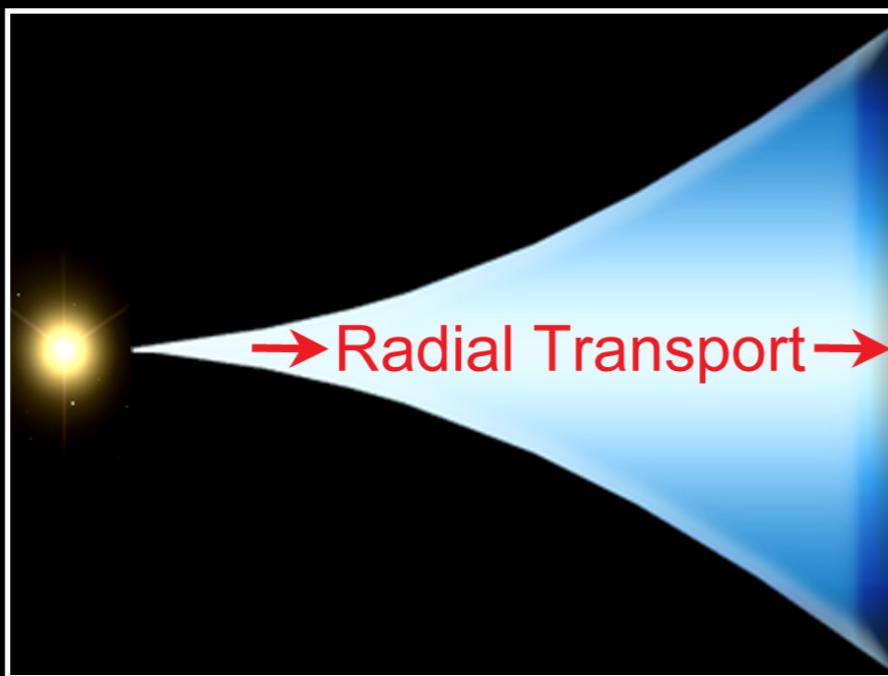
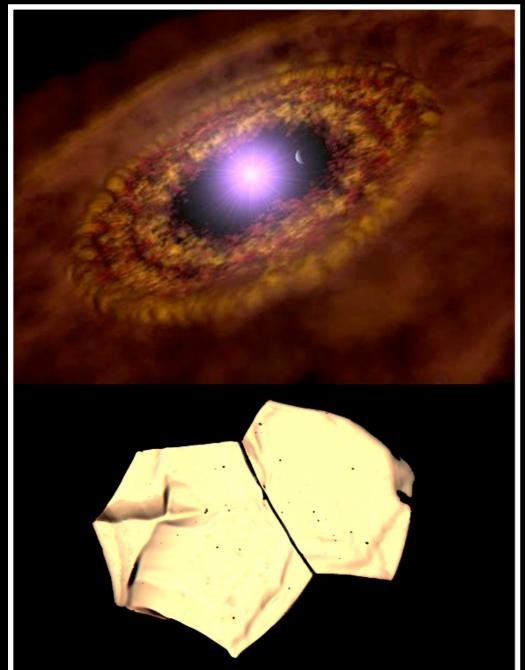
- ★ Nebular formation of troilite
- ★ Parent-body aqueous alteration & break-up
- ★ Delivery of cubanite to comet-forming region
- ★ Incorporation onto Wild 2



Genesis of Wild-2 Cubanite

Option 3

- ★ Nebular formation of troilite
- ★ Delivery of troilite to comet-forming region
- ★ Incorporation onto Wild 2
- ★ Cometary-body aqueous alteration



Characteristics of Stardust Sulfides
Fe, Ni, Cu, Zn, Mn-bearing
Mineralogy, Structure, Associations

Modified by capture into aerogel

Unmodified by capture into aerogel

Nebular Formation via gas-solid processes

Parent Body Precipitation

Aqueous Modification

on Wild 2

on other body

Implications for the genesis of Wild 2 sulfides, source material(s), and mixing in the early Solar System as indicated by compositions, structures, and uniformity/heterogeneity of phases.

Characteristics of Stardust Sulfides
Fe, Ni, Cu, Zn, Mn-bearing
Mineralogy, Structure, Associations

Modified by capture into aerogel

Unmodified by capture into aerogel

Nebular Formation of FeS via gas-solid processes

Parent Body Precipitation

Aqueous Modification

on Wild 2

on other body

Implications for large-scale solar system processes, including heat sources and aqueous alteration on a cometary body, as well as mechanisms and extent of radial mixing of material.

Cubanite...

- ★ *is the low-T form of CuFe₂S₃*
- ★ *is rare in the extraterrestrial collection*
- ★ *forms via low-T aqueous processes*

Stardust Cubanite...

- ★ *is evidence of radial mixing of sulfides from the inner Solar System to the comet-forming region*
 - ★ *mechanism & extent of mixing*
- ★ *likely formed *in situ* on Comet Wild 2*
 - ★ *heat sources for secondary processing*